

Targeting of HIV gp120 by oligonucleotide-photosensitizer conjugates

Light-induced damages

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Abstract Some guanine-rich oligonucleotides inhibit HIV infectivity through interaction with the gp120 glycoprotein. Besides, photoinactivation of viruses attracts attention for blood decontamination. The feasibility of targeting a red light-absorbing chlorin-type photosensitizer to gp120 through covalent coupling with 8-mer phosphodiester oligodeoxynucleotides is investigated. Some conjugates inhibit binding of antibodies directed to gp120. Inhibition is significantly increased upon red light activation. The activity of the conjugates correlates with their ability to self-associate, a process strongly favored by the propensity of the hydrophobic chlorin moiety to dimerize. Thus, the photosensitizer moiety both promotes structures with a higher affinity for gp120 and, upon light activation, can induce site-directed damages to the protein.

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Key words: HIV; gp120; Oligonucleotide; Photosensitizer; Chlorin; Self-association

1. Introduction

Recent advances in the knowledge of the mechanism of HIV entry into target cells [1] is opening new ways for prophylactic interventions. The entry is mediated through sequential binding of the HIV glycoprotein gp120 to the cellular CD4 receptor and co-receptors identified as the CXCR4 or CCR5 chemokine receptors, a choice depending on the virus tropism. The recent resolution of the gp120 crystal structure [2] allowed for the identification of the area involved in co-receptor recognition [3]. As surmised, it comprises the gp120 V3 loop that has been shown to be a major determinant of the virus tropism [4]. It is worth noting that the zone near the V3 loop base and the V3 loop itself which delineate the co-receptor binding site are highly positively charged [3,5,6]. Polyanions, such as sulfated polyesters, have long been recognized to inhibit HIV infection in vitro [7,8]. Among other polyanionic compounds, oligonucleotides have attracted attention. Indeed, aptamer strategies make it possible to select oligonucleotide sequences with high affinity for proteins that are not naturally bound by nucleic acids, which considerably extends the range of possible targets for oligonucleotides [9]. Thus, beyond the

expected charge effect presented by long homopolymers [10], shorter oligonucleotides with sequence-dependent anti-viral activity have been selected [11]. Although not an absolute requirement [12], the presence of G-tracts favoring the formation of folded structures or strand association involving G-tetrads led to the most efficient compounds [11,13,14]. A phosphorothioate backbone was also required [11]. In assays based on inhibition of HIV infection in vitro without any anticipation on the mechanism of action, gp120 and more specifically the V3 loop were identified as the main targets [11].

However, binding of oligonucleotides to their targets is a priori reversible. Composite molecules associating oligonucleotides with reactive substituents able to produce irreversible damages in the target are worth considering. In particular, photosensitizing substituents [15] that are not toxic in the dark sensitize singlet oxygen formation or react by electron transfer upon light excitation leading to chemical damages to molecules in the vicinity. We focused our attention on photosensitizers belonging to the chlorin family [16,17]. These molecules which absorb light above 620 nm can be activated in a region corresponding to minimal light absorption by blood, a prerequisite for photochemical blood sterilization [18]. We recently reported on the photoinactivation of various strains of the HIV-1 virus by a member of this family [19]. In the present study, we demonstrate the feasibility of targeting a hydrophobic chlorin-type photosensitizer to specific areas of the HIV envelope protein gp120, in particular the V3 loop, through coupling to some G-rich oligonucleotides.

2. Materials and methods

2.1. Chlorin-oligonucleotide conjugates

The chlorin-type photosensitizer referred to as CHEVP (Fig. 1) was prepared from the related vinyl porphyrin by a photochemical method [17]. It was coupled to different phosphodiester deoxyoligonucleotides derivatized at the 3'-position through a hexadamine linker and the conjugates were purified by high performance liquid chromatography (HPLC) as described elsewhere [15]. The absorption spectrum of the TTGGTTGG-chlorin conjugate is shown in Fig. 1.

2.2. Inhibition of antibody binding

Enzyme-linked immunosorbent assay (ELISA) binding assays were performed using 96 well microtiter plates (Immulon 4 from Dynatech). After the plates were coated with recombinant HIV-1 gp120_{IIIIB} (Intracel), oligonucleotides or chlorin conjugates in phosphate buffer saline, pH 7.4, were added to the wells at different concentrations (0, 1×10^{-7} , 1×10^{-6} and 1×10^{-5} M). After incubation (30 min), the plates were irradiated or kept in the dark for the same

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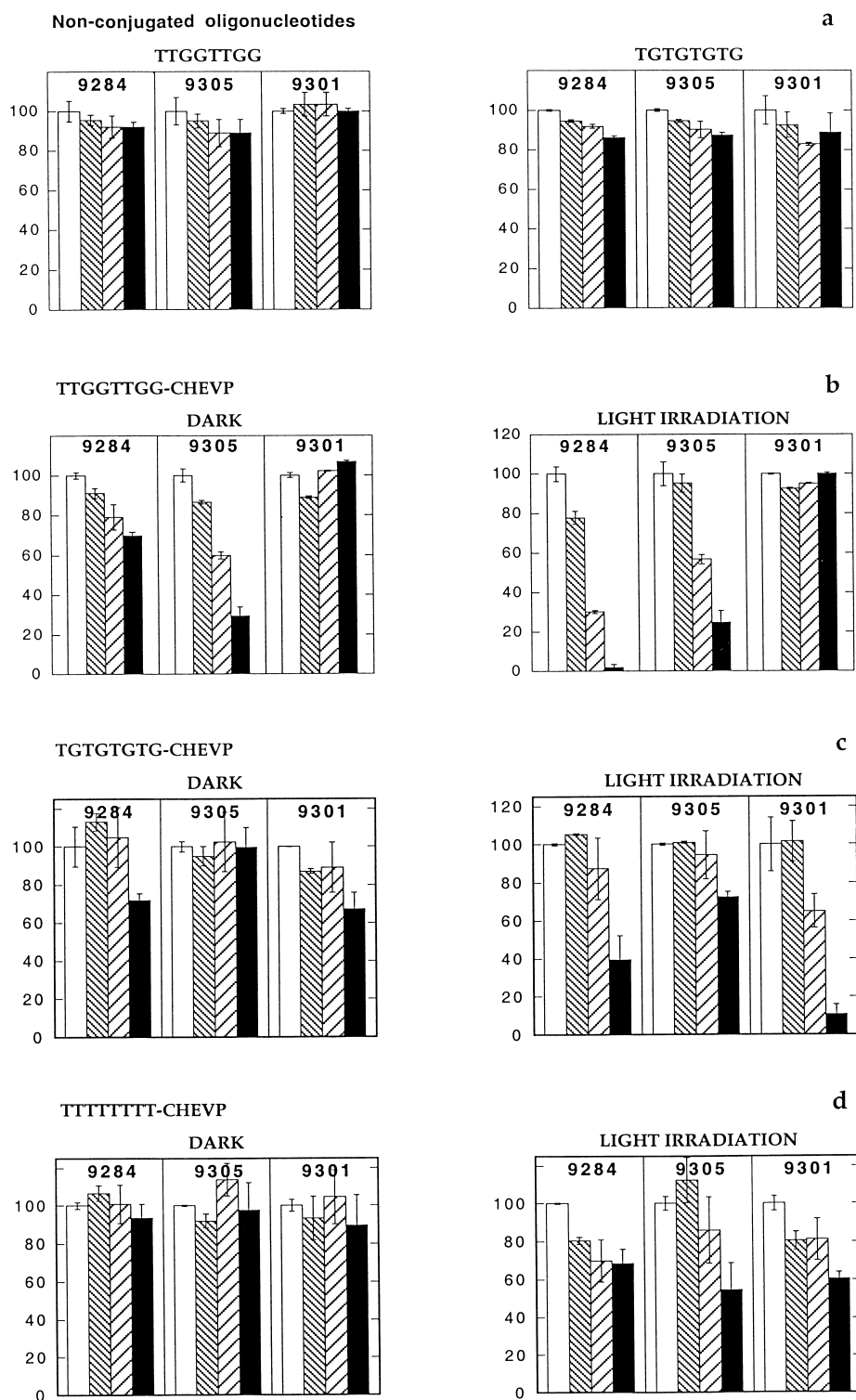


Fig. 2. Dose-dependent inhibition of the binding of gp120-directed antibodies by non-conjugated and chlorin-conjugated oligonucleotides. The antibodies are identified by the numbers 9284, 9305 and 9301 as indicated in the text. Experiments carried out in the dark or involving light irradiation are specified in the figure. In the case of the non-conjugated oligonucleotides (a), only experiments in the dark are shown, light irradiation having no effect, as expected. Experiments with TTGGTTGG-CHEVP, TGTGTGTG-CHEVP and TTTTTTTT-CHEVP are shown in b, c and d, respectively. Antibody binding was quantified by absorbance as described in Section 2, but in order to allow for comparison, data were normalized to 100 in the absence of competitors. The concentrations of the non-conjugated or chlorin-conjugated oligonucleotides are: open, 0; finely hatched, 1×10^{-7} M; broadly hatched, 1×10^{-6} M; filled, 1×10^{-5} M.

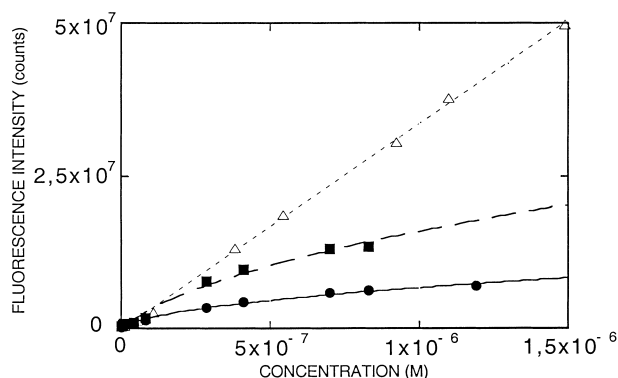


Fig. 3. Concentration-dependence of the fluorescence emission of TTTT-TT-CHEVP (Δ), TGTGTGTG-CHEVP ($\blacksquare$) and TTGGT-TGG-CHEVP ($\bullet$) in buffer solutions. The data are fitted according to Eq. 1 for a concentration-dependent dimerization process.

3.3. A dual role of the chlorin moiety

Besides its photochemical activity, it clearly appears that the chlorin moiety is required to observe interactions with gp120, even in the dark. Due to its lack of solubility in water, control experiments with the non-conjugated chlorin were precluded, unfortunately. Two hypotheses can be put forward, both based on the hydrophobic character of this molecule. First, it can be anticipated that the chlorin moiety interacts with hydrophobic loci on the protein. Such sites have been identified at the apex of the V3 loop [25] and near its base [2]. A conjunction of hydrophobic and positively charged loci can also be found around the epitope of mAb 9301. Secondly, the chlorin moiety could favor the formation of particular structures owing to its ability to self-associate through π - π interactions.

The second hypothesis was further investigated by monitoring the concentration-dependence of the fluorescence emission spectra of the conjugated oligonucleotides in buffer solutions or in solutions containing a detergent, Triton X-100, able to prevent association. Good linearity was observed for all the conjugates in Triton solutions, indicating that they remain monomeric. As shown in Fig. 3, this is no more true for aqueous solutions. Also, a strong effect of the oligonucleotide sequence is observed. An excellent linearity is seen for the TTTT-TT-chlorin conjugate, indicating that the presence of the chlorin moiety is not sufficient by itself to promote association. Departure from linearity is clearly seen for the G-containing conjugates. A non-linear concentration-dependence of the fluorescence of hydrophobic porphyrins or chlorins has been observed by several authors and was attributed to dimerization [20]. The shape and the wavelength maxima of the emission and the excitation spectra of the conjugates were found to be independent of the concentration, indicating that only monomers fluoresce. The changes in fluorescence upon concentration were treated accordingly to derive dimerization constants [20]. Theoretical fits assuming dimerization are shown in Fig. 3. The data were not fitted properly assuming other hypotheses such as tetramer formation. The dimerization constant for the TTGGT-TGG-chlorin conjugate, $K = (3.3 \pm 0.8) \times 10^7 \text{ M}^{-1}$, was more than one order of magnitude higher than that measured for the TGTGTGTG-chlorin conjugate, $K = (2.4 \pm 0.5) \times 10^6 \text{ M}^{-1}$. The higher stability of the TTGGT-TGG-chlorin dimer might be due to the forma-

tion of two stacked guanine quartets involving two oligonucleotide strands folded in a hairpin structure [26]. In the case of the oligonucleotide with alternating TG, although the formation of at least one guanine quartet is still possible upon folding as hairpin and dimerization, the absence of two contiguous guanine plateaus would lead to a less stable structure. Taken together, these results indicate that both interactions between chlorin moieties and formation of guanine quartets contribute to the conjugate association. Further characterization of these structures by melting analysis using UV spectroscopy or fluorescence was made difficult by a marked irreversibility. This was apparently due to sticking of the conjugate on the optical cell walls when the temperature was raised. This phenomenon probably arose upon strand dissociation and unfolding exposing the chlorin moiety that could become more prone to adsorption onto the optical glass.

In summary, upon covalent linking with a hydrophobic photosensitizer, some oligonucleotides become able to inhibit binding of antibodies specific of the HIV gp120 glycoprotein. It is worth noting that derivatization of other G-rich oligonucleotides by hydrophobic residues also confers upon them anti-HIV activity [27]. The hydrophobic moiety may promote strand association leading to structures with specific protein affinity. Association is well demonstrated in the case of the planar chlorin macrocycle used in the present study. The stability of the dimer formed also depends on the sequence of the oligonucleotide part and strongly suggests an important contribution of the formation of guanine quartets. The association of conjugates as dimers is a major determinant of activity as illustrated by the 'control' T₈-chlorin conjugate. The comparison of the two G-rich chlorin-oligonucleotide conjugates investigated here shows that the oligonucleotide sequence also comes into play in targeting these molecules to specific regions of gp120. Then, site-directed photochemical damages can be elicited upon irradiation of the conjugates associated to gp120 using red light, which significantly increases their efficiency. In addition to their potential as new anti-viral agents for blood treatment, such conjugates might find use as biochemical tools to explore protein functions. In particular, their ability to produce permanent damages could help in elucidating the role played by protein areas that are temporarily exposed [28].

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